
Quantum chemical study of adsorption at metal oxide surfaces

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This thesis presents a Quantum Chemical Study of Adsorption at Metal Oxide Surfaces. Ab initio molecular electronic structure calculations have been carried out using the GAUSSIAN 98 code to examine: (i) Transition metal atoms on oxide supports density functional calculations (ii) CO adsorption on Ni, Pd, Cu and Ag on MgO, CaO, SrO and BaO: Density functional Calculations. The thesis consists of Three chapters scheduled as follow: CHAPTER 1 This chapter presents a literature survey and a review of theory: Density Functional Theory (DFT) as well as the methods and simulation techniques employed in the calculations. CHAPTER 2 In this chapter, The adsorption of the transition metal atoms Cu, Ni, Ag, and Pd on F and F⁺ defect sites as well as on regular O²⁻ positions of the MgO, CaO, SrO, and BaO (001) surfaces with the help of DFT calculations employing cluster models has studied. The model clusters were embedded in simulated Coulomb fields that closely approximate the Madelung fields of the host surfaces, and the adsorption properties have been analyzed with reference to the basicity and energy gap of the oxide support in addition to orbital interactions. The following conclusions have been extracted: • We found that adsorption properties of the F site, which contains a pair of electrons, is reminiscent of the properties of the regular O²⁻