Study the effect of addition of some metals chlorides in nanosize on the Physical Properties of polyvinyl alcohol (PVA)

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The aim of the present work is to prepare nanocomposite polymer samples based on metal chlorides and polyvinyl alcohol (PVA) for the electronic applications. PVA/CoCl2, PVA/NiCl2 and PVA/MnCl2have been prepared by means of in situ synthesis via the solvent casttechnique. The prepared nanocomposite polymer samples withconcentrations (41.6, 32.2, 19.2, 4.5 wt%) of CoCl2, NiCl2 and 16.5, wt%) of MnCl2 have been characterized 3.8 varioustechniques, X-ray diffraction (XRD), the differential scanningcalorimetric (DSC) and the scanning electron microscope (SEM). DCand AC conductivities are -examined at different temperatures andfrequencies; also optical absorption in UV Visible range was studied. The growth of metal chloride crystallites in the polymer matrix innanoscale for samples of higher contents of metal chlorides has beenobserved Using XRD and SEM studies. Furthermore, the intensity of PVA characterizing peak has decreased while its broadness has beenincreased which revealed that the amorphisity of the polymer matrix hasincreased. The temperature dependence of DC electrical conductivity for allinvestigated samples has been studied between 303 and 443 K. Theresults showed that the electrical conduction is thermal activated processobeying Arrhenius relation. The activation energy was calculated andfound in the range from 0.64 to 0.26 eV for all investigated samples. The AC conductivity (σ ac), dielectric constant (ϵ '), dielectric loss(ϵ '') and loss tangent ($tan\delta$) were performed in the temperature range 303-443 K over the frequency range 100 Hz-100 KHz. The frequencyAbstractdependence of the AC conductivity shows that oac increases withincreasing frequency for all investigated samples. The AC conductivity obeyed the ωs power law. The values of the power s of all samples havebeen extracted at different temperatures which suggest the andtunneling conduction mechanisms. frequency The temperaturedependence of the dielectric constant (ϵ '), dielectric loss (ϵ '') and losstangent (tan \square) have been studied. The general behavior showed that ϵ', ϵ'' and tan [] decreased with increasing frequency for all investigatedsamples obeying Debye dispersion equations. The dependence of the dielectric constant (ϵ') on temperature shows two broad peaks related to a relaxation processes. The optical absorption spectra were performed in the wavelengthrang (200-900 nm) for all mentioned samples. The absorptionmechanism is due to allowed direct and indirect transitions with oneoptical band gap for composites contained MnCl2 and two band

gaps forcomposites contained CoCl2 and NiCl2. Co+2 ions are present in mixedoctahedral and tetrahedral form while Ni+2 ions are present in octahedralform in the polymeric matrix. The optical band gap, absorption edge and Urbach energy were evaluated and their values change with the changein both particle size of metal chlorides and their concentrations insidethe polymer matrix.